No direct relationship was concluded between activity and reducibility of the catalysts. The higher activity of supported lanthanum manganite is attributed to the higher surface area of catalyst, small particles of $LaMnO_3$ and subsequently the 2-propanol molecules can easily access to active sites of catalyst and react with oxygen. $LaMnO_3$-ZSM-5 showed higher activity, even than 1% Pt/Al_2O_3. The study revealed that the supporting of the perovskite-type catalyst on the zeolites improved the activity of the catalyst. $LaMnO_3$-ZSM-5 exhibited the higher activity and stability than $LaMnO_3$ in catalytic combustion of 2-propanol revealing the superior role of ZSM-5 support in improving the activity of the catalyst. $LaMnO_3$-ZSM-5 could be a promising catalyst in environmental control applications.

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