Table 5. Comparison result of Cl_3CCO_2H with the reported catalysts in literature for the synthesis of highly functionalized dihydro-2-oxypyrroles **4i** and **4l**.

Entry	Product	Catalyst	Conditions	Time (h)	Yield (%)	Ref.
1	**4i**	Cl_3CCO_2H	MeOH, r.t.	4	95	This work
2		$Cu(OAc)_2 \cdot H_2O$	MeOH, r.t.	5	85	[18]
3		I_2	MeOH, r.t.	1	81	[16]
4		AcOH	EtOH/ 70 °C	4	85	[15]
5		Oxalic acid dehydrate	MeOH, r.t.	2	87	[28]
6		[Hpyro][HSO_4]	MeOH, r.t.	6	80	[22]
7		$Al(H2PO4)_3$	MeOH, r.t.	5	80	[23]
8		$InCl_3$	MeOH, r.t.	3	85	[27]
9		[n-Bu_4N][HSO_4]	MeOH, r.t.	4	86	[24]
10	**4l**	Cl_3CCO_2H	MeOH, r.t.	4	91	This work
11		$Cu(OAc)_2 \cdot H_2O$	MeOH, r.t.	6	91	[18]
12		I_2	MeOH, r.t.	1	82	[16]
13		AcOH	EtOH/ 70 °C	-	-	[15]
14		Oxalic acid dehydrate	MeOH, r.t.	2	89	[28]
15		[Hpyro][HSO_4]	MeOH, r.t.	6	82	[22]
16		$Al(H2PO4)_3$	MeOH, r.t.	5	81	[23]
17		$InCl_3$	MeOH, r.t.	3	85	[27]
18		[n-Bu_4N][HSO_4]	MeOH, r.t.	4	88	[24]

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