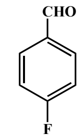
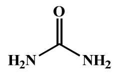
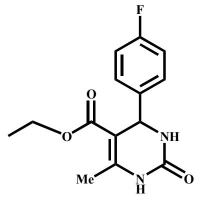
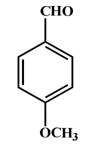
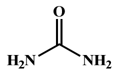
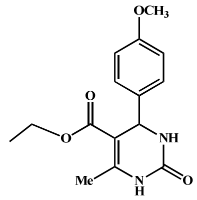
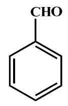
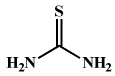
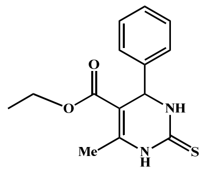
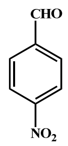
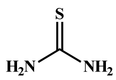
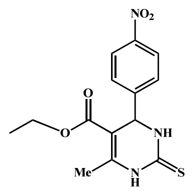
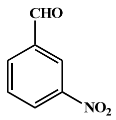
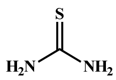
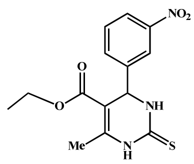
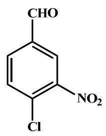
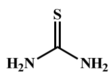
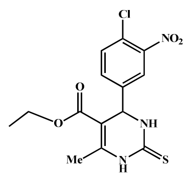
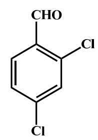
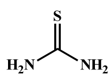
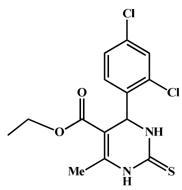
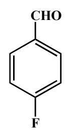
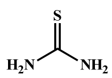
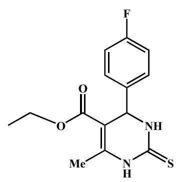
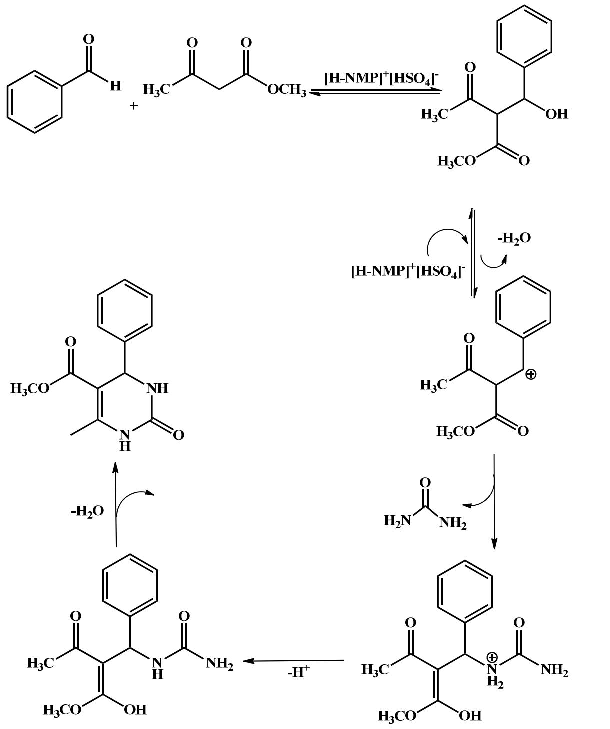
Table 4. *(Continued).*

7				5	94
8				12	92
9				9	90
10				8	94
11				11	93
12				7	95
13				10	93
14				8	90

[a]Isolated yields.



Scheme 1. The proposed Knoevenagel reaction mechanisms for synthesis of dihydropyrimidinones.

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Table 5. The synthesis of 4c using different catalysts.

Entry	Catalyst	Time (min)	Yield (%)	Ref.
1	Cu@PMO-IL, 60 °C, solvent-free	120	80	[32]
2	TSILs, 100 °C, solvent-free	10	88	[33]
3	[Cbmim] Cl, Reflux, MeCN	90	69	[34]
4	ZnO, 80 °C, BMI·BF4	12 h	67	[35]
5	MWCNTs-OSO_3H, 120 °C, solvent-free	20	94	[11]
6	Al-MCM-41, 80 °C, solvent-free	15	84	[12]
7	$H_6GeW_{10}V_2O_{40}.22H_2O$, 80 °C, solvent-free	3 h	91	[13]
8	$[H\text{-}NMP]^+[CH_3SO_3]^-$, 90 °C, solvent-free	4	97	This work

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