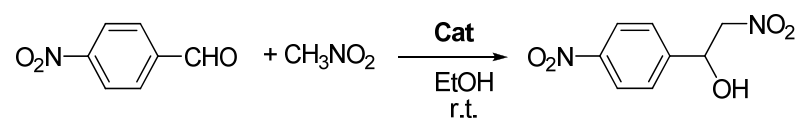


Scheme 2. Proposed catalytic cycle for the diastereoselective formation of β -nitroalcohols catalyzed by **Cat-2**.

Table 4. Yield of the model reaction with the recycling of **Cat-1** and **Cat-2**.



Run	Yield (%) ^a	
	Cat-1	Cat-2
1	98	97
2	97	96
3	96	96
4	93	95
5	91	93
6	70	92
7	63	92
8	58	88

^aIsolated yield.

Conclusion

In summary, an organic superbase (*N,N*-dimethyl biguanide), immobilized on mesoporous silica SBA-15 and magnetically separable nanoparticle supports was found, for the first time, to act as efficient catalyst for the Henry reaction under mild conditions. Employing these new nanocatalysts, the reactions were carried out in fast (5-10 min), clean and cost-effective manner with possible reusability of the catalysts for several runs. Avoiding of organic volatile solvents and toxic catalysts enabled an operationally green new portal to β -nitroalcohols. In the case of magnetite support, ease in separation of the catalyst with an external magnet simply eliminates the filtration process and make the condition even greener. Additionally, there was a good stereoselectivity in the formation of β -nitroalcohols obtained from the reaction of nitroethane with aldehyde which was believed due to the dual activation role of the catalyst.

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